

# The University of Bradford Institutional Repository

<http://bradscholars.brad.ac.uk>

This work is made available online in accordance with publisher policies. Please refer to the repository record for this item and our Policy Document available from the repository home page for further information.

To see the final version of this work please visit the publisher's website. Access to the published online version may require a subscription.

**Link to publisher's version:** <http://dx.doi.org/10.1021/ja510338r>

**Citation:** Pitto-Barry A, Barry NPE, Russo V et al (2014) Designing supramolecular liquid-crystalline hybrids from pyrenyl-containing dendrimers and arene ruthenium metallacycles. Journal of the American Chemical Society. 136(50): 17616-17625.

**Copyright statement:** This document is the Accepted Manuscript version of a Published Work that appeared in final form in the Journal of the American Society, © 2014 American Chemical Society after peer review and technical editing by the publisher. To access the final edited and published work see <http://dx.doi.org/10.1021/ja510338r>

# Liquid-Crystalline Properties of Pyrenyl-Containing Dendrimers Encapsulated in Arene Ruthenium Metalla-Cycle

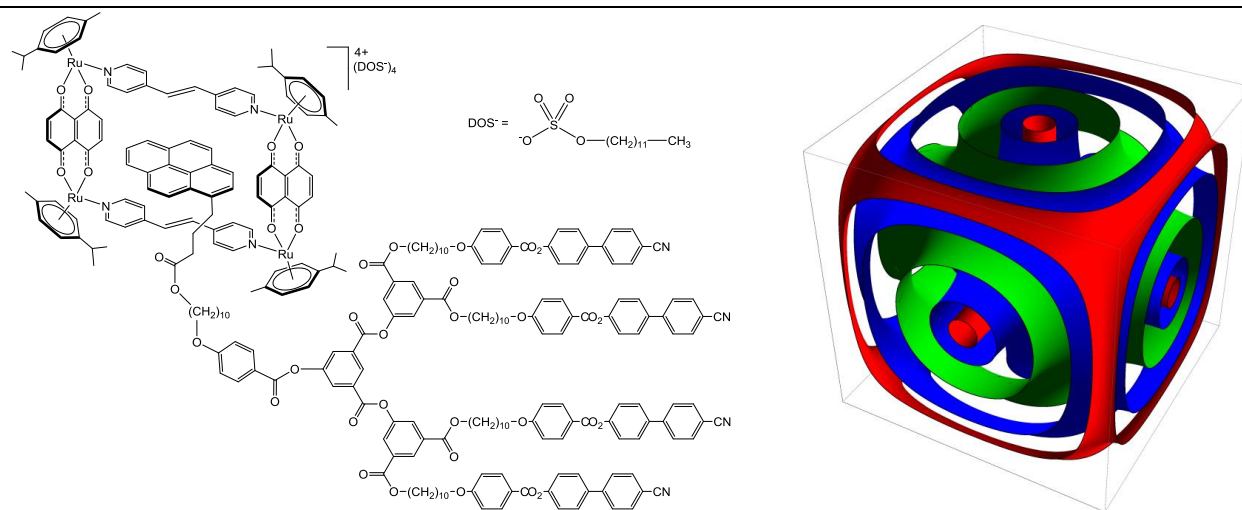
Anaïs Pitto-Barry,<sup>†</sup> Nicolas P. E. Barry,<sup>†</sup> Virginie Russo,<sup>†</sup> Benoît Heinrich,<sup>§</sup> Bertrand Donnio,<sup>\*,§,¶</sup> Bruno Therrien,<sup>\*,†</sup> and Robert Deschenaux<sup>\*,†</sup>

<sup>†</sup>Institut de Chimie, Université de Neuchâtel, Avenue de Bellevaux 51, 2000 Neuchâtel, Switzerland

<sup>§</sup>Institut de Physique et Chimie des Matériaux de Strasbourg (IPCMS), UMR 7504, CNRS-Université de Strasbourg, 23 rue du Loess, BP43, 67034 Strasbourg cedex 2, France

<sup>¶</sup>Complex Assemblies of Soft Matter Laboratory (COMPASS), UMI 3254, CNRS/Solvay/University of Pennsylvania, CRTB, 350 George Patterson Boulevard, Bristol, PA 19007, USA

Supporting Information Placeholder



**ABSTRACT:** The association of the arene ruthenium metalla-cycle [Ru<sub>4</sub>(p-cymene)<sub>4</sub>(bpe)<sub>2</sub>(donq)<sub>2</sub>][DOS]<sub>4</sub> (bpe = 1,2-bis(4-pyridyl)ethylene, donq = 5,8-dioxydo-1,4-naphthoquinonato, DOS = dodecyl sulfate) with pyrenyl-functionalized poly(arylester) dendrimers bearing cyanobiphenyl end-groups is reported. The supramolecular dendritic systems display mesomorphic properties as revealed by polarized optical microscopy, differential scanning calorimetry and small-angle X-ray scattering investigations. The multi-component nature of the dendrimers and of the corresponding host-guest supramolecules (i.e. end-group mesogens, dendritic core, pyrene unit, aliphatic spacers and metallacycle) leads to the formation of highly segregated mesophases with a complex multi-layered structure owing to the tendency of the various constitutive building-blocks to separate in different organized zones. The pyrenyl dendrimers exhibit a multilayered smectic A-like phase, thereafter referred to as LamSmA phase to emphasize this unaccustomed morphology. As for the corresponding Ru<sub>4</sub>-metalla-cycle adducts, they self-organize into a multi-continuous thermotropic cubic phase with the *Im* $\bar{3}m$  space group symmetry. This represents a unique example of liquid-crystalline behavior observed for such large and complex supramolecular host-guest assemblies. Models of their supramolecular organizations within both mesophases are proposed.

## INTRODUCTION

Self-organization of metallomesogens (metal-containing liquid crystals) is of interest to combine the properties of both the metal ion (e.g., optical, electronic, magnetic, catalytic properties, coordination geometry) and the mesogenic part (e.g., structure-dependent organization, fluidity, stimuli-responsive

behavior).<sup>1</sup> Incorporation of metallic ions in liquid-crystalline (LC) materials extends further possible novel intermolecular interactions such as metal-metal (metallophilic) and metal-ligand interactions which can have important consequences on the physical properties (e.g., amplification or emergence of new properties) and on the self-assembling abilities.<sup>1</sup> A wide

variety of metallomesogens has been generated to date using concepts based on coordination, organometallic and supramolecular chemistry. Such a diversity of structures embraces from simple to very sophisticated molecular systems, including one-metal center systems,<sup>2</sup> metal dimer molecules,<sup>3</sup> metal-metal bonded molecules,<sup>4</sup> metallocenes, buckyferrocenes,<sup>5</sup> metallo-helicates,<sup>6</sup> metallacrowns,<sup>7</sup> clustomesogens,<sup>8</sup> heteropolymetallic metallomesogens,<sup>9</sup> metallomesogenic polymers and two-dimensional networks.<sup>10</sup> Mesogens can also be introduced in metal-containing liquid crystals as counter-ions.<sup>11</sup> Incorporation of metal-containing building-blocks in liquid-crystalline molecules results usually in substantial modifications of the initial mesomorphic properties of the ligands (e.g., changes of mesophase symmetry, thermodynamic stability, transition temperatures). When these blocks are much larger than the mean size of the mesogenic part, mesomorphism can even be suppressed. However, their incorporation into LC phases has been successfully achieved through the use of voluminous mesomorphic dendritic systems,<sup>12</sup> and beautiful examples of LC catenanes<sup>13</sup> and rotaxanes<sup>14</sup> among other systems<sup>15-16</sup> can be found in the literature.

Herein, we report the synthesis, properties and supramolecular organization of the first members of a new family of liquid-crystalline host-guest complexes based on an organometallic Ru<sub>4</sub> cage (host) and liquid-crystalline dendrimers (guests). The synthesis of the third-generation (P<sub>3</sub>) poly(arylester) dendrimers containing cyanobiphenyl mesogens functionalized with a pyrenyl unit is reported (Supporting Information). The preparation of the second-generation dendrimer P<sub>2</sub> was already described (Supporting Information). The synthesis of the tetranuclear arene ruthenium salt [Ru<sub>4</sub>(*p*-cymene)<sub>4</sub>(bpe)<sub>2</sub>(donq)<sub>2</sub>][DOS]<sub>4</sub> (bpe = 1,2-bis(4-pyridyl)ethylene; donq = 5,8-dioxydo-1,4-naphthoquinonato; DOS = dodecyl sulfate), abbreviated as [1][DOS]<sub>4</sub>, is also reported. The cationic metalla-cycle [1]<sup>4+</sup> was previously isolated as its triflate salt,<sup>17</sup> but no liquid-crystalline behavior was observed when P<sub>2</sub> or P<sub>3</sub> was included in [1][CF<sub>3</sub>SO<sub>3</sub>]<sub>4</sub>. Clearly, the choice of the counter-ion is crucial for the design of metallomesogens.<sup>1,18</sup> Therefore, in this study, we used the dodecyl sulfate anion (DOS), a long and flexible ionic surfactant, which usually contributes efficiently to the decrease of the melting points of the corresponding salts.<sup>18</sup>

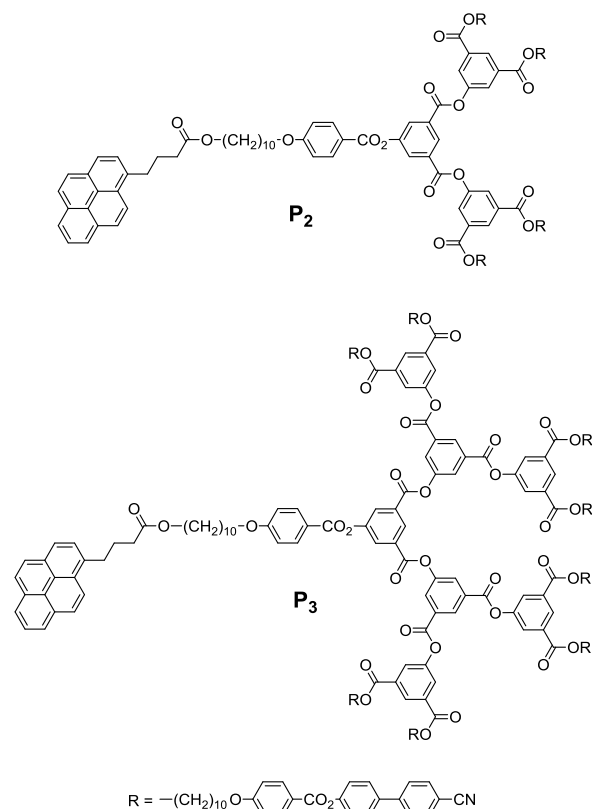
The liquid-crystalline properties and supramolecular organization of the pyrenyl-based dendrimers P<sub>n</sub> and their inclusion complexes [P<sub>n</sub>⊂1][DOS]<sub>4</sub> were investigated by polarized optical microscopy (POM), differential scanning calorimetry (DSC) and small-angle X-ray scattering (SAXS). The dendrimers showed a multilayered smectic A-like phase (referred thereafter to as LamSmA) in agreement with their multiblock architecture and the presence of peripheral calamitic mesogens. The [P<sub>n</sub>⊂1][DOS]<sub>4</sub> revealed an unexpected supramolecular organization into a multi-continuous cubic phase.

## RESULTS AND DISCUSSION

**Design.** The assembly of 1 with P<sub>n</sub> is based on the ability of the cage to interact with the planar  $\pi$ -conjugated pyrenyl moiety via hydrophobic and  $\pi$  -  $\pi$  interactions.<sup>17</sup> Second- and third-generation dendrimers were selected as liquid-crystalline promoters to thwart the size of the cage. Among the tetra-, hexa- and octacationic arene ruthenium metalla-cycles already known,<sup>19</sup> we selected the species with the lowest global

charge. This choice was motivated by previous experiments in which anti-cancer drug candidates composed of liquid-crystalline dendrimers included in a hexacationic arene ruthenium metalla-prism, [Ru<sub>6</sub>(*p*-cymene)<sub>6</sub>(tpt)<sub>2</sub>(donq)<sub>3</sub>]<sup>6+</sup> (tpt = 2,4,6-tri(pyridine-4-yl)-1,3,5-triazine) were found to be non-mesomorphic.<sup>20</sup>

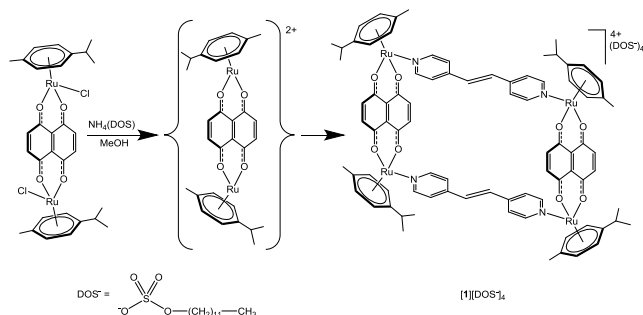
**Synthesis of the dendrimers.** Second- and third-generation pyrenyl poly(arylester) dendrimers, P<sub>2</sub> and P<sub>3</sub>, respectively (Figure 1), were synthesized by two successive esterification reactions using [*N,N'*-dicyclohexylcarbodiimide (DCC), 4-(dimethylamino)pyridinium *para*-toluenesulfonate (DPTS)] as coupling Agents.<sup>21</sup> The first esterification reaction between 1-pyrenebutyric acid and 4-((10-hydroxydecyl)oxy)benzoic acid<sup>22</sup> resulted in the formation of a new pyrene derivative, which was subsequently esterified with the corresponding second- and third-generation phenol-based dendrons<sup>22</sup> (Scheme S1). Dendrimers P<sub>2</sub> and P<sub>3</sub> were characterized by <sup>1</sup>H, <sup>13</sup>C NMR, UV-visible and IR spectroscopies, elemental analysis and ESI-mass spectrometry (Supporting Information).



**Figure 1.** Structures of the second- and third-generation dendrimers P<sub>2</sub> and P<sub>3</sub>.

**Synthesis of the arene ruthenium metalla-cycle.** The tetranuclear arene ruthenium complex [Ru<sub>4</sub>(*p*-cymene)<sub>4</sub>(bpe)<sub>2</sub>(donq)<sub>2</sub>][DOS]<sub>4</sub> ([1][DOS]<sub>4</sub>) was prepared in methanol from the dinuclear complex [Ru<sub>2</sub>(*p*-cymene)<sub>2</sub>(donq)<sub>2</sub>Cl<sub>2</sub>] and the bidentate ligand bpe in the presence of ammonium dodecyl sulfate in good yield (73%) following a procedure already used to obtain this cationic rectangle as a triflate salt (Scheme 1).<sup>17</sup> Here, the ammonium cation is used as a halide scavenger.

**Scheme 1. Synthesis of arene ruthenium metalla-cycle isolated as its dodecyl sulfate salt, [1][DOS]<sub>4</sub> (conditions: MeOH, NH<sub>4</sub>DOS, bpe, 60 °C, 24h).**

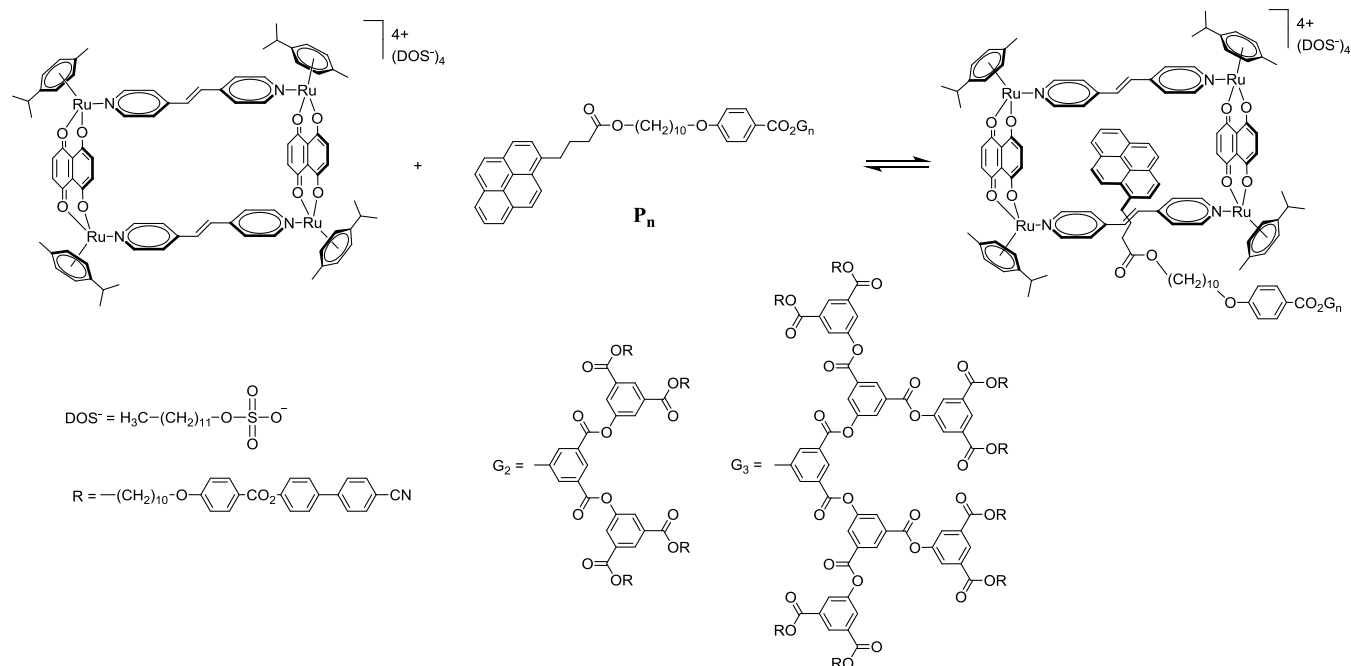


The <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 25 °C) of [1][DOS]<sub>4</sub> is in agreement with the spectrum of its triflate analogue [1][triflate]<sub>4</sub>.<sup>17</sup> The signals of the bpe protons are shifted up-field as compared to those of the free bpe. The signal of the 5,8-dioxydo-1,4-naphthoquinonato bridging ligands is shifted downfield as compared to the parent complex [Ru<sub>2</sub>(*p*-cymene)<sub>2</sub>(donq)<sub>2</sub>Cl<sub>2</sub>], while the methyl, isopropyl and phenyl resonances of the *p*-cymene protons are almost unchanged. The signals of the aliphatic protons of the dodecyl sulfate counter-ion are well resolved and found at 3.76, 1.46-1.34 and 0.72 ppm, respectively. The <sup>13</sup>C NMR spectrum of the salt was

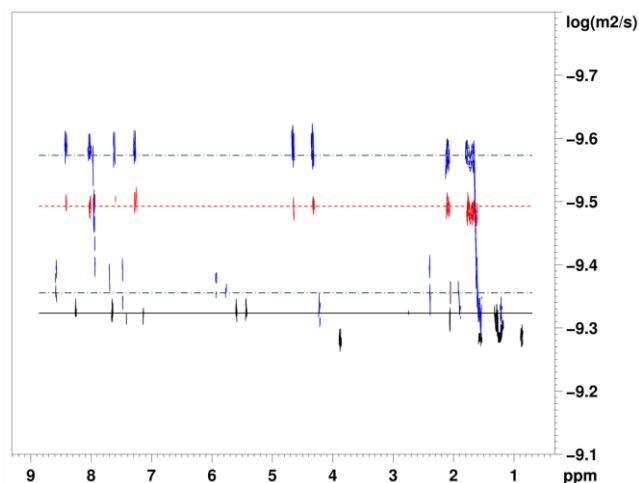
recorded in CD<sub>2</sub>Cl<sub>2</sub> at 25 °C and fully attributed (Supporting Information). Metalla-cycle [1][DOS]<sub>4</sub> is not stable under ESI-mass spectrometry but the fragmentation peak corresponding to [Ru<sub>2</sub>(*p*-cym)<sub>2</sub>(donq) + bpe + (DOS)]<sup>+</sup> is found at 1107.27. This fragmentation is typical of arene ruthenium metalla-cycles.<sup>23</sup> Despite being unstable under ESI-mass spectrometry, such metalla-cycles have been found to be extremely stable in solution and even at elevated temperature.<sup>23</sup> The purity of [1][DOS]<sub>4</sub> was confirmed by elemental analysis (Supporting Information).

**Synthesis of the supramolecular assemblies.** A solution of one equivalent of both **P<sub>n</sub>** and [1][DOS]<sub>4</sub> was stirred during two days in CH<sub>2</sub>Cl<sub>2</sub> at ambient temperature (Scheme 2). Then, the solvent was removed by slow evaporation. The compounds were dried for several days. Elemental analysis confirmed the purity of the systems and the absence of residual solvent (Supporting Information). Note that direct grinding of one equivalent of [1][DOS]<sub>4</sub> with one equivalent of **P<sub>2</sub>** or **P<sub>3</sub>** did not lead to materials displaying liquid-crystalline properties. In this case, heterogeneous black areas of non-melted metalla-cycle with zones of melted dendrimer (in the mesophase) were observed by POM. This observation indicates that the interactions in solution between the pyrenyl part and the metalla-cycle are crucial for the formation of homogenous liquid-crystalline systems.

**Scheme 2. Synthesis of the [P<sub>n</sub>⊂1][DOS]<sub>4</sub> assemblies (conditions: CH<sub>2</sub>Cl<sub>2</sub>, RT, 2 days).**



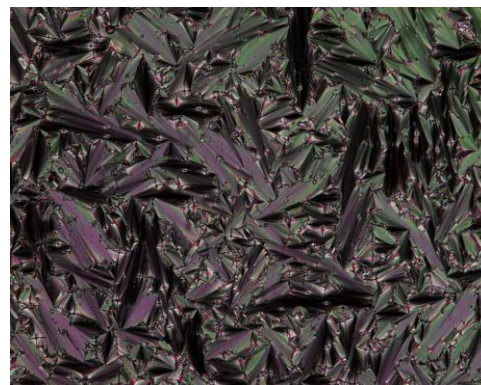
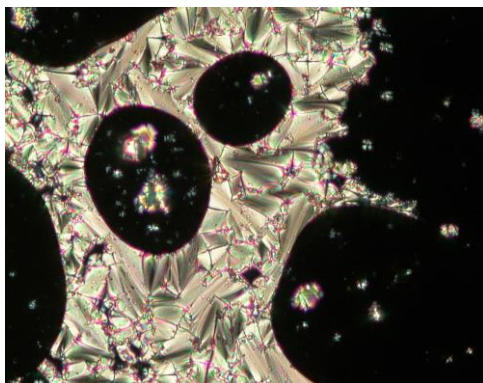
each part of the assembly when they are mixed together in solution, which indicates strong host-guest interactions.



**Figure 2.** DOSY NMR spectra of **P**<sub>3</sub> (red), [1]<sup>4+</sup> (black) and the corresponding associated system (blue) in CD<sub>2</sub>Cl<sub>2</sub> at 21 °C.

The <sup>13</sup>C NMR spectrum of the two systems was recorded in CD<sub>2</sub>Cl<sub>2</sub> at 25 °C and fully attributed (Supporting Information). The ESI-mass spectra of the two associated systems display the same fragmentation pattern [Ru<sub>2</sub>(*p*-cym)<sub>2</sub>(donq) + bpe + (DOS)]<sup>+</sup> at 1107.27 than the spectrum of [1][DOS]<sub>4</sub>. The compounds were also characterized by UV-Vis and IR spectroscopy (Supporting Information).

**Liquid-crystalline properties and supramolecular organization of dendrimers **P**<sub>2</sub> and **P**<sub>3</sub>.** The two pyrenyl-functionalized dendrimers **P**<sub>n</sub> led to the formation of a broad SmA-like phase above their glass transition temperatures (*T*<sub>g</sub>) (Table 1). The mesophases were partly identified by POM from the formation of typical focal-fan conic textures and homeotropic areas (Figure 3), indicative of un-tilted lamellae. The increase of the clearing temperature is concomitant with the increase of the dendritic generation as usually observed with this family of dendrimers.<sup>16,25</sup> This trend confirms that the stability of the mesophases increases with the number of mesogens.



**Figure 3.** Thermal-polarized optical micrographs of the focal-conic fan textures of the SmA-like phase displayed by **P**<sub>2</sub>, at T = 150 °C (top) and **P**<sub>3</sub>, at T = 145 °C (bottom) upon cooling the samples from the isotropic liquid.

The structures of the mesophases displayed by **P**<sub>2</sub> and **P**<sub>3</sub> were studied by SAXS. The samples were measured from ambient temperature up to 160 °C (**P**<sub>2</sub>) and 200 °C (**P**<sub>3</sub>). Analysis of the diffractograms (Figures 4, 5 and S6) confirmed that both dendrimers **P**<sub>2</sub> and **P**<sub>3</sub> display a single smectic-like phase, in agreement with POM observations.

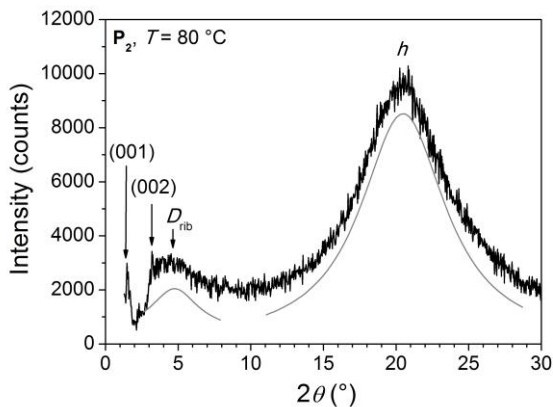
**Table 1. Phase-transition temperatures of dendrimers **P**<sub>2</sub> and **P**<sub>3</sub>.**

Compound	<i>T</i> <sub>g</sub> <sup>a</sup>	Transition <sup>b</sup>	<i>T</i> <sub>i</sub> <sup>a</sup>	Δ <i>H</i> (kJ·mol <sup>-1</sup> )
<b>P</b> <sub>2</sub>	33	LamSmA→I	165	9.4
<b>P</b> <sub>3</sub>	60	LamSmA→I	219	39.6

<sup>a</sup>*T*<sub>g</sub> = glass transition temperature, *T*<sub>i</sub> = isotropization temperature. <sup>b</sup>LamSmA = multilayered smectic A-like phase (see text), I = isotropic liquid. Temperatures (in °C) are given as the onset of the peak obtained during the second heating run; the *T*<sub>g</sub> were determined during the first cooling cycle (rate: 10 °C·min<sup>-1</sup>).

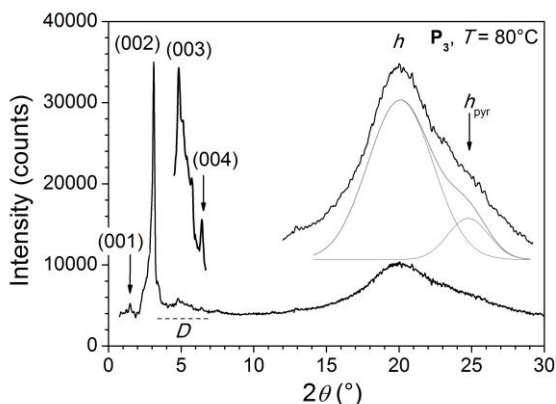
For **P**<sub>2</sub>, two sharp low-intensity reflections in a 1:2 ratio were detected in the small-angle range at 53.8 and 26.7 Å (Figure 4, reflections indexed as (001) and (002), respectively), confirming the layered structure of the mesophase with a periodicity of *d* = 53.8 Å. In addition, were also observed the classical wide-angle broad scattering with a maximum at ca. 4.4-4.6 Å originating from the overlapping distances between molten aliphatic chains (*h*<sub>ch</sub>), dendritic core (*h*<sub>dend</sub>) and mesogens (*h*<sub>mes</sub>), respectively (Figure 4, *h* = *h*<sub>ch</sub> + *h*<sub>mes</sub> + *h*<sub>dend</sub>), and another large, small-angle diffusion with a maximum at ca. 21-22 Å which was attributed to some molecular correlations caused by specific short-range ordering of the pyrene segments (labeled as *D*<sub>rib</sub>, Figures 4 and S6) within the layers (see below).





**Figure 4.** Representative SAXS patterns of **P<sub>2</sub>**

Above 40–50°C, the X-ray diffraction patterns of **P<sub>3</sub>** (Figure 5) display four sharp small-angle reflections in the 1:2:3:4 ratio [at 55.5, 27.5, 18.4 and 13.8 Å, indexed as (00*l*) reflections and *l* = 1–4], which also attest for the formation of a lamellar phase with a periodicity of  $d = 55.2$  Å ( $d = \sum l \cdot d_{00l} / N_l$ , where  $N_l$  is the number of 00*l* reflections). As for **P<sub>2</sub>**, similar broad scattering signals ( $h = 4.4$ – $4.6$  Å and  $D = 25$ – $26$  Å, the latter slightly shifted to smaller angles) are also observed for this compound (Figure 5 and S6). In addition, one can also recognize a broad signal at *ca.* 3.5 Å,  $h_{\text{pyr}}$ , attributed to  $\pi$  stacking of pyrenyl cores (see below).



**Figure 5.** Representative SAXS patterns of **P<sub>3</sub>**

X-ray patterns thus unambiguously confirm the formation of a smectic-like phase for **P<sub>2</sub>** and **P<sub>3</sub>**. As the generation is increased, the presence of up to four thin reflections indicates good molecular segregation (sharp interfaces) of the different constitutive blocks of the dendrimers. Thus, the mesogenic groups connected at the periphery of the dendrimers act as classical calamitic smectogens, and therefore direct the formation of a lamellar phase, similarly to other structurally related dendrimers.<sup>12,16</sup> The specific multi-block molecular architecture of **P<sub>n</sub>** dendrimers further imposes the segregation between the cyanobiphenyl units and the appended pyrene with the dendritic core and aliphatic spacers.<sup>12</sup> This is clearly evidenced by the abnormal intensity profile distribution of the low-angle reflections series, with a relatively exhausted first-order (001) reflection and an enhanced second-order (002) reflection revealing the periodic alternation of several high-electronic density sub-layers associated to the mesogenic end-groups and the dendritic scaffold, respectively, with low-

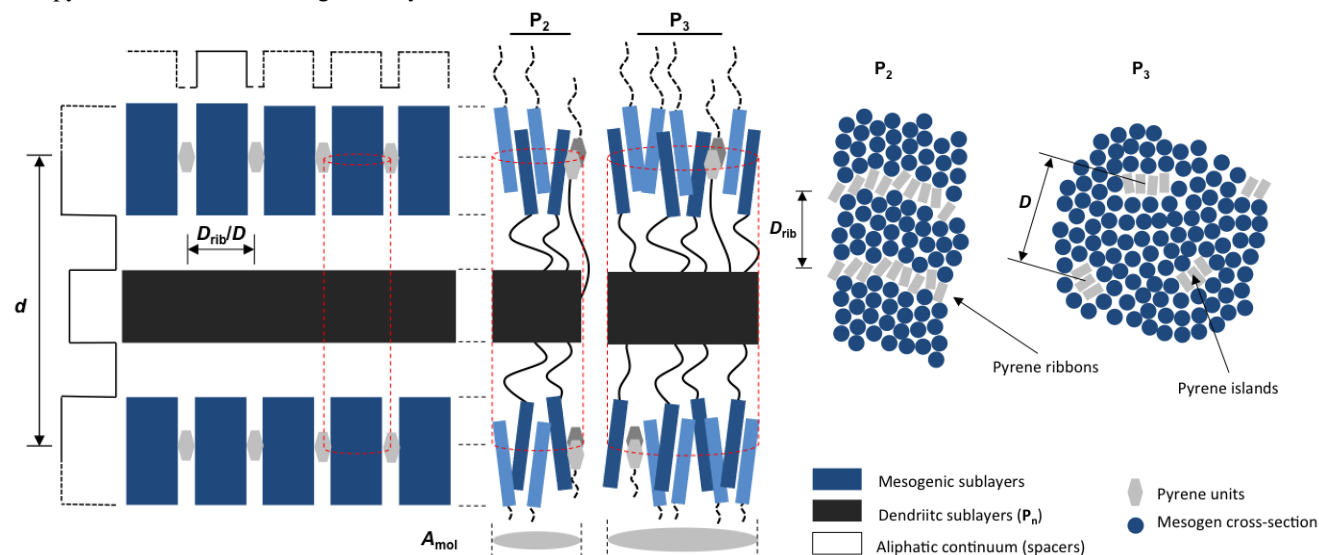
electronic density sub-layers associated to the molten aliphatic spacers (Figures 4, 5 and S6). The observation of the characteristic optical texture of the SmA confirms this organization for the mesogens, and the disordered liquid-like state of the dendritic cores within the median sublayer. Apart from the formation of the lamellae, a further segregation process occurs inside the mesogen sublayers between the cyanobiphenyl mesogens and the pyrene groups, as revealed by the appearance of the additional small-angle scattering signal (labeled  $D_{\text{rib}}$  and  $D$ , for **P<sub>2</sub>** and **P<sub>3</sub>**, respectively). The long spacer between the pyrene moiety and the dendron allows the former to reach the sub-layer formed by the mesogenic groups. The relative high intensity of this signal arises from the insertion of a fraction of the spacer segments in the mesogenic sublayer, due to the different end-group lengths (*ca.* 18 and 8 Å for mesogen and pyrene, respectively). However, the patterning of the layer perforation remains short-range, as deduced from the broad signal shape (correlation lengths from Scherrer formula are about 25 and 50 Å for **P<sub>2</sub>** and **P<sub>3</sub>**). The peak location  $D_{\text{rib}} \approx 21$  Å for the lower generation is compatible with the segregation of pyrene and spacer segments in continuous ribbons alternating with strips formed by the 4 times more abundant cyanobiphenyl mesogens fraction (Figure 6). The continuity of the ribbons can however not be maintained in the higher generation compound because of the further dilution of the pyrene moieties, as confirmed by the maximum shift to  $D \approx 26$  Å in **P<sub>3</sub>**: pyrene and spacer segments then segregate into islands positioned at the nodes of a loose pseudo-hexagonal lattice. The average aggregation number of associated pyrene groups can be estimated from  $D$  by using an approach detailed elsewhere,<sup>26</sup> giving 3–4 units per island (Figure 6). The patterning of the layer optimizes the separation in space of incompatible mesogens and incorporated spacer segments, but still interferes with the parallel alignment of the mesogens and with the cohesion of the layer. The reduced hindrance with the smaller islands indeed constitutes one feature explaining the higher isotropization temperature and the larger isotropization enthalpy per mesogen of **P<sub>3</sub>**.

For a more detailed understanding of the mesophase structure, the lamellar packing can be further characterized by the molecular area,  $A_{\text{mol}}$ , ratio of the molecular volume,  $V_{\text{mol}}$ , and layer spacing,  $d$ . For **P<sub>2</sub>** and **P<sub>3</sub>**, the molecular volumes are estimated to be around *ca.* 4970 and 9240 Å<sup>3</sup> respectively (considering a density close to 0.95 at 80°C) and the molecular areas *ca.* 92 and 167 Å<sup>2</sup>, respectively, quasi doubling from **P<sub>2</sub>** to **P<sub>3</sub>** in consistency with the doubling of the number of peripheral cyanobiphenyl units (4 → 8) and generation (2 → 3). These values and the quasi-invariance of the lamellar periodicities ( $d = 53.8$  Å versus  $d = 55.2$  Å) whilst the volume is roughly doubled as the generation increases traduce the molecular expansion within the layers and a similar packing mode for both systems, which evidently is imposed by the unfolded dendritic core. These  $A_{\text{mol}}$  values are nevertheless significantly smaller than the overall cross section of the mesogen end-groups (estimated to be 115 and 205 Å<sup>2</sup> from reference volumetric data).<sup>27</sup> Such a discrepancy is commonly encountered for mesogens with one end devoid of terminal chain, as mesogens attached to segments from neighboring sublayers can either associate tip-to-tip into dimers or interdigitate over their whole length.<sup>27</sup> Depending upon the statistical proportion of dimers, the mesogens can either form monolayer, bilayer, or partial-bilayer arrangements (Figure S7).<sup>28</sup> This dimer propor-

tion lays here at 35%, as typically found for cyanobiphenyl compounds.

On the basis of these calculations and the relative intensities of the reflections, the structure of the lamellar phase consists of a central sub-layer containing the expanded, disordered dendritic cores, separated from the perforated mesogenic layers (from the pyrenes and intercalated chain fractions) by the aliphatic continuum (made from all the spacers) (Figure 6). In both cases, the dendrimers logically adopt an elongated pseudo "cylindrical" conformation with the peripheral mesogens and pyrenes distributed homogeneously on either side of the

dendritic scaffold, and separated by the aliphatic spacers. Such a highly segregated molecular conformation is in agreement with the fundamental lamellar periodicities. Since this structure differs substantially from a classical SmA phase, this phase was preferentially labeled as LamSmA to highlight its particular multilayer structure. Such lamellar mesophases with sublayers patterned by ribbons ( $P_2$ ) or islands ( $P_3$ ) thus best accommodate the multisegregated architecture. Despite of being rarely reported, they have nevertheless been previously observed in block copolymers and small molecules with rod-like mesogens bearing side-on substituents.<sup>29</sup>

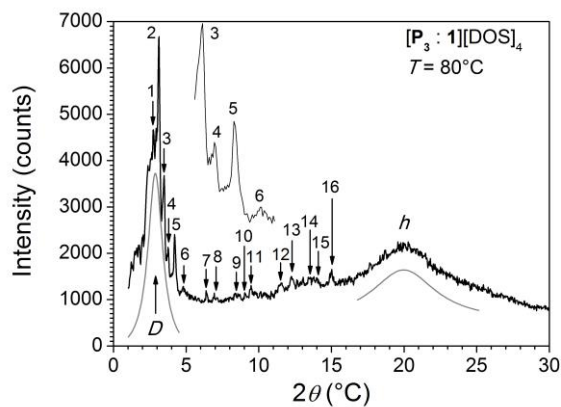


**Figure 6.** Postulated schematic supramolecular organization of  $P_2$  and  $P_3$  within the LamSmA phase; the side view (left) shows the alternation of the different sublayers, while the top view (right) reveals the local perforations of the mesogenic outlayers. Profiles (far left): electronic density modulations in the direction perpendicular to the layers ( $d$ ) and within the layers ( $D_{rib}/D$ ).

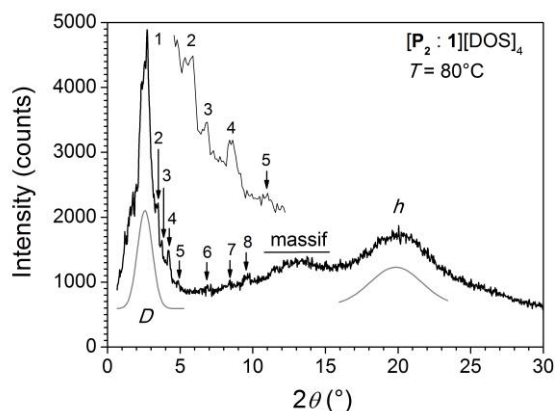
**Liquid-crystalline properties and supramolecular organization of the  $[P_2C1][DOS]_4$  and  $[P_3C1][DOS]_4$  host-guest complexes.** In their molten state, none of the complexes appeared birefringent when observed by POM, but only the formation of highly viscous, large dark, texture-less areas were detected when pressure was applied on the glass slide. Above 100°C, degradation was observed resulting likelihood from the irreversible dissociation of the supramolecular species, a situation often encountered in liquid-crystalline inclusion complexes.<sup>30</sup> The structure of their mesophases below 100°C was eventually deduced by SAXS, namely a thermotropic cubic phase (with the  $Im\bar{3}m$  space group symmetry), from ambient-temperature up to *ca.* 100°C for  $[P_2C1][DOS]_4$ , and between 50°C and *ca.* 100°C for  $[P_3C1][DOS]_4$  (below 50°C, an amorphous state is observed). No practical information was withdrawn from DSC since dissociation/decomposition of the samples occurs before the clearing point was reached.

Both systems exhibit similar X-ray patterns (Figures 7 and 8) with several small-angle reflections and two large diffuse scatterings, one at wide-angle evidencing the molten state of the chains, dendritic cores and mesogens ( $h = 4.4\text{--}4.5$  Å), and another one in the small-angle range (at *ca.* 32–33 Å), which is attributed to short-range correlated structures emerging from the distribution of different zones. Indeed, this periodicity,  $D$ , is related to the ionic lattice and arises from the electron-rich metalla-cycle, and from the segregation patterning with the cyanobiphenyl end-groups, (presumed to be similar to the perforations in the LamSmA phases for the  $P_n$  guests).

The SAXS pattern of  $[P_3C1][DOS]_4$  reveals 16 sharp, small-angle reflections (Figure 7, Table S1) with reciprocal  $d$ -spacings in ratios theoretically compatible with both primitive ( $P$ ) and body-centered ( $I$ ) space groups, whilst it totally excludes a face-centered  $cfc$  network ( $F$ ). The SAXS pattern of  $[P_2C1][DOS]_4$  recorded at 80°C (Figure 8, Table S2) is similar, but exhibits fewer reflections, with a series of 8 defined peaks, and a massif of unresolved wide-angle signals. The lower number of reflections *a priori* precludes a definitive phase assignment for  $[P_2C1][DOS]_4$ , but the similarity between the SAXS patterns of both host-guest complexes suggests that they both self-organize in the same type of cubic phase.



**Figure 7.** Representative SAXS patterns of  $[P_3<1][DOS]_4$



**Figure 8.** Representative SAXS patterns of  $[P_2<1][DOS]_4$

Although the diffraction patterns were theoretically compatible with a primitive lattice ( $P$ ), the allowed groups ( $Pm\bar{3}$ ,  $Pn\bar{3}$ ,  $Pm\bar{3}m$ ,  $Pm\bar{3}n$ ,  $Pn\bar{3}n$  and  $Pn\bar{3}m$ ) leave the absence of too many reflections unexplained and can be safely excluded. Just two centered space groups can be considered,  $Im\bar{3}$  and  $Im\bar{3}m$ , since the groups reflection conditions are satisfied.<sup>31</sup> Aggregation into the highest symmetry is generally admitted, and, accordingly, the  $Im\bar{3}m$  space group ( $N^\circ 229$ ) was retained.<sup>31</sup> Therefore, the symmetry of the cubic phases of the  $[P_n<1][DOS]_4$  host-guest systems is characterized by a body-centered cubic network ( $I$ ), with lattice parameters  $a = 85.4$  and  $86.8$  Å for  $[P_2<1][DOS]_4$  and  $[P_3<1][DOS]_4$ , respectively (Tables S1 and S2). Based on the partial volume of  $[1][DOS]_4$  estimated to  $3600$  Å<sup>3</sup>, and thus considering approximated molecular volumes,  $V_{mol}$ , of  $8600$  and  $12800$  Å<sup>3</sup> (density,  $\rho \approx 1.05$  g·cm<sup>-3</sup> at  $80^\circ\text{C}$ ) for both complexes, each cubic unit cell contains *ca.*  $72 \pm 2$  and  $50 \pm 2$  supramolecules, respectively. These large aggregation numbers combined with the shape of the SAXS patterns accord with a highly segregated, multi-continuous structure (see below). This is moreover unusual that the observable reflections are so numerous and propagate so far in the mid-angle range for thermotropic cubic phases (Tables S1 and S2). This feature proves the high degree of nano-segregation between incompatible segments, as they are confined in domains separated by sharp interfaces. The insertion of the metalla-cycle DOS salt certainly preserves the differentiation of dendritic and mesogen moieties inherited from  $P_2$

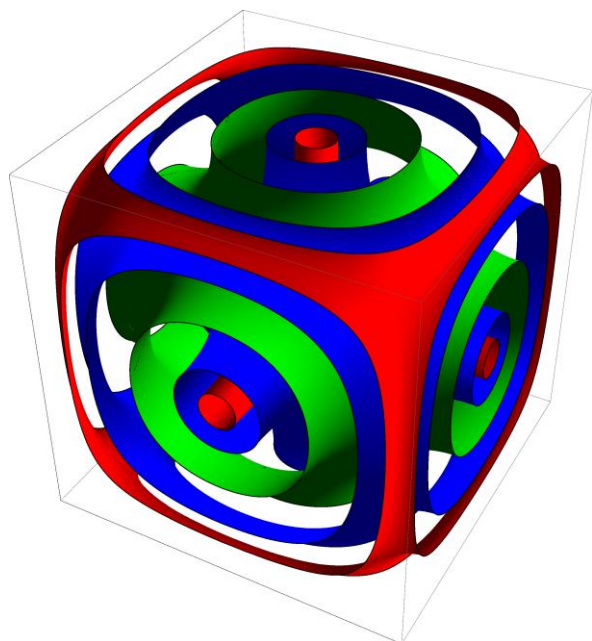
and  $P_3$  pyrenyl dendrimers (see below), but concomitantly imposes intricate and sterically constrained arrangements of the corresponding segregated domains. Nearly the same intricate multi-continuous structure is obtained for both host-guest systems, with even an extension of the reflection series to larger Miller indices for the higher-generation dendrimer, following the trend observed in the lamellar phase of the pyrenyl dendrimers.

Molecular organizations proposed for previous thermotropic multi-continuous cubic phases are based on minimal surfaces of type  $P$  and  $G$ , in accordance with the observed  $Im\bar{3}m$  and  $Ia\bar{3}d$  symmetries.<sup>32,33</sup> As seen above, the  $Ia\bar{3}d$  symmetry, and therefore the  $G$  minimal surfaces, are here clearly excluded [presence of the (310), (222), (330) / (441) reflections being incompatible with this group], and the  $Im\bar{3}m$  symmetry confirmed. Remarkably, some authorized reflections are missing in the series, as the ones at lowest angles [i.e. (110), (200), (211)]. Although the overlapping with the intense small-angle scattering may partially hinders the detection of these reflections, their absence (or weakness) follows an intensity modulation that goes through a maximum for reflections (310) and (222) in the generation 3, and for reflection (220) in the generation 2. As a regular decrease is commonly observed for bi-continuous phases,<sup>33</sup> this modulation is a further proof for a more complex multi-continuous structure.<sup>34,35</sup>

The emergence of the three-dimensional structure evidently follows the expansion of the mesogen and aliphatic sublayers by the metalla-cycle DOS salt, what creates discrepancies with the molecular area of the dendritic-containing sublayer. Such a distortion of the layering is compensated by the curvature at both interfaces (i.e. mesogenic/aliphatic and dendritic/aliphatic interfaces) and results in the fusion of the lamellae into intricate, intermingled three-dimensional networks (interwoven “labyrinths” in Figure 9). Tentatively, a more comprehensive description of the organization is proposed through the adaptation of a multi-continuous structural model, already validated for some thermotropic  $Im\bar{3}m$  systems.<sup>33</sup> This model is based on the foliation of the lattice space by the various interfaces on either sides of the  $P$ -type infinite, periodic minimal surface (IPMS) of zero mean average curvature (materialized by the green isosurface in Figure 9). The cyanobiphenyl-containing layers swollen by ionic metalla-cycles wrap this minimal surface. Simultaneously, the dendritic moieties are rejected either along the edges and intersections at the corners, or along lines going through the centers of the faces and of the cube, and clustered into interwoven labyrinth-like domains. The inhomogeneities of the latter zones (e.g. shrinkage on edges and dilation on vertices) are modulated by the great flexibility and deformability of the dendritic cores. The fluid aliphatic continuums are intercalated between both these two types of disjointed domains, and delimited respectively by the interfaces with the mesogenic parts (blue isosurfaces in Figure 9) and by the somewhat shrunken interfaces with the dendritic core (red interfaces in Figure 9). This multicontinuous model is therefore highly reminiscent of the lamellar structure of the  $P_n$  guests, which shows the same sequence of strata from the minimal surface toward edges and vertices. A geometric link between both structures is found with the half-face diagonals, which in the model is normal to the successive segment strata (see below), and those lengths ( $\sim 61$  Å) are close to the layer spacing. A perfect coincidence was indeed not expected, due to several concomitant effects, among which the contribution of the metalla-cycles, the curvature of the interfaces and the



related shrinking of the strata with the distance to the minimal surface.

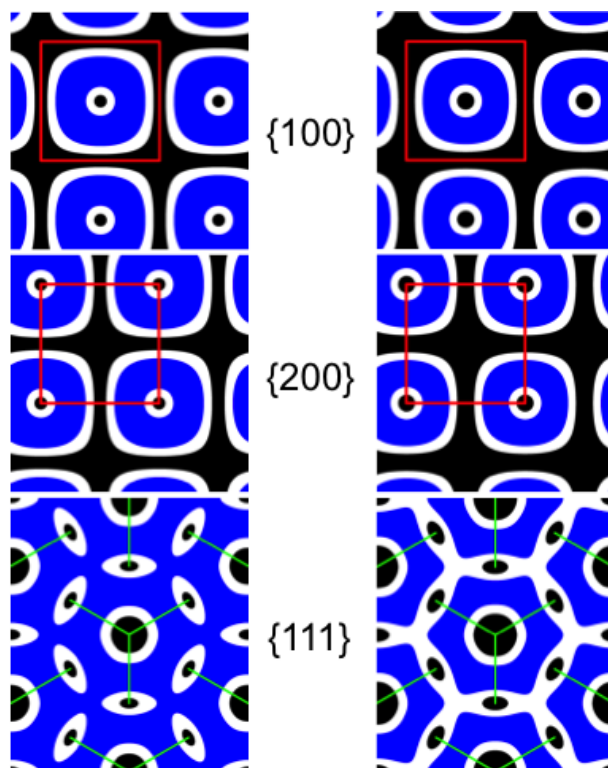


**Figure 9.** Schematic representation of the multi-continuous cubic structure (surface foliation model, references 33, 36, 37) for the host-guest complex  $[P_2C1][DOS]_4$ . Representation of the different intermediate interfaces  $\beta$  (red for the aliphatic/dendritic interfaces,  $|\beta| = 1.89$ ; blue for the mesogen/aliphatic interfaces,  $|\beta| = 1.44$ ) on both sides of the isosurface of zero-mean curvature ( $\beta = 0$ , green).

The portion of the stratum covered by a single segment block, i.e. the area per segment, compensates to a certain extent the wedge shape of the entire encapsulated molecule. These areas become then closer to the natural cross-sections and reduce the interfaces between the various incompatible segments for an overall improved packing. The efficiency of the process can be quantified using surface foliations of the lattice volume,<sup>36,37</sup> whereby the isosurfaces sharing the lattice in accordance with the segment volume fractions are assimilated into interfaces. For both molecules, this geometric modeling yields areas per segment<sup>38</sup> that decrease from *ca.* 235-240 and 345-350 Å<sup>2</sup>, respectively, on the minimal surface, to 190-195 and 295-300 Å<sup>2</sup>, at the interface with the aliphatic strata and finally to 120-125 Å<sup>2</sup> and 215-220 Å<sup>2</sup> at the interface with the dendritic part. The latter areas are in agreement with the expected overall cross-section of the dendritic branches (roughly 110 and 200 Å<sup>2</sup>) while the larger areas close to the minimal surface are compatible with the contribution of inserted ionic moieties. Remarkably, even the cross-sections of aliphatic segments are approached at both interfaces, when taken into account the intermingling DOS chain ends.

The surface foliation model accounts for an almost ideal compensation of the cross section discrepancies and is so far consistent with all experimental data, unlike a few other reported systems, though of completely different nature.<sup>35</sup> For a clearer insight of the distribution of segregated zones in the cubic cell, sectional simulated sketches within most important crystallographic planes are represented for both complexes (Figure 10). The families of planes  $\{100\}$  and  $\{200\}$  contain the two labyrinths' directions and cut the third orthogonally. The strata

crossed by the half-face diagonals remind the lamellae of the pyrenyl-dendrimers (see above). The  $\{111\}$  plane cuts all three labyrinths' directions with the same angle. This plane might evolve toward a columnar lattice plane at higher swelling degrees, when edges resorb and superposed vertices merge to stratified columns. Instead of swelling mesogens, the shrinking of the dendritic part might yield a further original "mixed micellar-bicontinuous" mesophase, when edges stay wrapped by the aliphatic strata, while the dendritic fractions aggregate on the vertices.



**Figure 10.** Simulated distribution of segregated zones in the cubic cell for complexes  $[P_2C1][DOS]_4$ , (left), and  $[P_3C1][DOS]_4$  (right), within most important crystallographic families of planes (blue: mesogens+metalla-cycles sublayers; white: aliphatic strata; black: dendritic cores labyrinths; red frame: cubic lattice; green lines: edges to vertices from next plane).

The general design of these materials offers a wide range of possibilities to modulate the mesophases' structures. Further deepening of these structures would require additional experimental information, such as more accurate volumes and shapes of segments, probably in combination with systematic variations of spacer and mesogen lengths. Such future developments would also be desirable due to the original molecular architecture, different from mesogens connected to unequal aliphatic tails as previously described in the literature.

## CONCLUSION

The syntheses of pyrenyl-functionalized poly(arylester) dendrimers with cyanobiphenyl end-groups and of a tetranuclear arene ruthenium metalla-cycle  $[Ru_4(p\text{-cymene})_4(bpe)_2(donq)_2]^{4+}$  isolated as a dodecyl sulfate salt are reported. The assembly of these two entities leads to supramolecular organometallic compounds. The dendrimers show a liquid-crystalline behavior, which is identified as a multi-

layered smectic (LamSmA) phase, whereas the supramolecular assemblies either do not display mesomorphic properties or self-organize into an intricate, multicontinuous  $Im\bar{3}m$  cubic phase, when isolated as triflate and dodecylsulfate salts, respectively. The wedge shape of the dodecylsulfate assemblies seems to be a key condition for the appearance of this mesophase, for which a structural model was proposed.

## ASSOCIATED CONTENT

**Supporting Information.** Synthesis and characterization of all new compounds. Elucidation of the mesophase from the diffraction pattern is described. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## AUTHOR INFORMATION

### Corresponding Author

bertrand.donnio@ipcms.unistra.fr  
bruno.therrien@unine.ch  
robert.deschenaux@unine.ch

### Funding Sources

R.D. thanks the Swiss National Science Foundation (Grant No 200020-140298) for financial support.

## ACKNOWLEDGMENT

A generous loan of ruthenium chloride hydrate from Johnson Matthey Technology Centre is gratefully acknowledged. B.D. and B.H. are very grateful to Dr Guillaume Weick (IPCMS) for his help regarding the IPMS mathematical calculations and to the CNRS for support.

## REFERENCES

- (1) (a) Donnio, B.; Guillon, D.; Bruce, D. W.; Deschenaux, R., *Metallomesogens*, In *Comprehensive Coordination Chemistry II: From Biology to Nanotechnology*. McCleverty, J. A.; Meyer, T. J. Eds. Elsevier: Oxford, UK, 2003. Volume 7 (Fujita, M.; Powell, A.; Creutz, C. eds), pp. 357-627. (b) Donnio, B.; Guillon, D.; Bruce, D. W.; Deschenaux, R., *Metallomesogens*. In *Comprehensive Organometallic Chemistry III: From Fundamentals to Applications*. Crabtree, R. H.; Mingos, D. M. P. Eds. Elsevier: Oxford, UK, 2006. Volume 12 (D. O'Hare ed.), pp. 195-294. (c) Pucci, D.; Donnio, B., Metal-containing liquid crystals. In *Handbook of Liquid Crystals*. Edited by Goodby, J. W.; Collings, P. J.; Kato, T.; Tschierske, C.; Gleeson, H.; Raynes, P. Wiley-VCH, Weinheim, **2014**, Volume 5.
- (2) (a) Date, R. W.; Fernandez Iglesias, E.; Rowe, K. E.; Elliott, J. M.; Bruce, D. W. *Dalton Trans.* **2003**, 1914-1931. (b) Binnemans, K. *J. Mater. Chem.* **2009**, *19*, 448-453.
- (3) (a) Rao, N. V. S.; Choudhury, T. D.; Paul, M. K.; Francis, T. *Liq. Cryst.* **2009**, *36*, 409-423. (b) Prokhorov, A. M.; Santoro, A.; Williams, J. A. G.; Bruce, D. W. *Angew. Chem. Int. Ed.* **2012**, *51*, 95-98.
- (4) (a) Frein, S.; Auzias, M.; Sondenecker, A.; Vieille-Petit, L.; Guintchin, B.; Maringa, N.; Süß-Fink, G.; Barberá, J.; Deschenaux, R. *Chem. Mater.* **2008**, *20*, 1340-1343. (b) Santoro, A.; Wegrzyn, M.; Whitwood, A. C.; Donnio, B.; Bruce, D. W. *J. Am. Chem. Soc.* **2010**, *132*, 10689-10691.
- (5) (a) Kim, D.-J.; Oh, N.-K.; Lee, M.; Choi, M.-G. *Mol. Cryst. Liq. Cryst.* **1996**, *280*, 129-134. (b) Massiot, P.; Impéror-Clerc, M.; Veber, M.; Deschenaux, R. *Chem. Mater.* **2005**, *17*, 1946-1951. (c) Matsuo, Y.; Muramatsu, A.; Kamikawa, Y.; Kato, T.; Nakamura, E. *J. Am. Chem. Soc.*, **2006**, *128*, 9586-9587.
- (6) (a) El-ghayoury, A.; Douce, L.; Skoulios, A.; Ziessel, R. *Angew. Chem. Int. Ed.* **1998**, *37*, 2205-2208. (b) Ziessel, R.; Douce, L.; El-ghayoury, A.; Harriman, A.; Skoulios, A. *Angew. Chem. Int. Ed.* **2000**, *39*, 1489-1493.
- (7) (a) Barberá, J.; Elduque, A.; Giménez, R.; Lahoz, F. J.; López, J. A.; Oro, L. A.; Serrano, J. L. *Inorg. Chem.* **1998**, *37*, 2960-2967. (b) Kishimura, A.; Yamashita, T.; Yamaguchi, K.; Aida, T. *Nature Mater.* **2005**, *4*, 546-549. (c) Frischmann, P. D.; Guieu, S.; Tabeshi, R.; MacLachlan, M. J. *J. Am. Chem. Soc.* **2010**, *132*, 7668-7675.
- (8) (a) Terazzi, E.; Bourgogne, C.; Welter, R.; Gallani, J.-L.; Guillon, D.; Rogez, G.; Donnio, B. *Angew. Chem. Int. Ed.* **2008**, *47*, 490-495. (b) Molard, Y.; Dorson, F.; Cîrcu, V.; Roisnel, T.; Artzner, F.; Cordier, S. *Angew. Chem. Int. Ed.* **2010**, *49*, 3351-3355. (c) Moicanu, A. S.; Amela-Cortes, M.; Molard, Y.; Cîrcu, V.; Cordier, S. *Chem. Commun.* **2011**, *47*, 2056-2058. (d) Terazzi, E.; Rogez, G.; Gallani, J.-L.; Donnio, B. *J. Am. Chem. Soc.* **2013**, *135*, 2708-2722.
- (9) Kadkin, O.; Galyametdinov, Y.; Rakhmatullin, A. *Mol. Cryst. Liq. Cryst.* **1999**, *332*, 2619-2628.
- (10) (a) Deschenaux, R.; Turpin, F.; Guillon, D. *Macromolecules* **1997**, *30*, 3759-3765. (b) Deschenaux, R.; Jauslin, I.; Scholten, U.; Turpin, F.; Guillon, D.; Heinrich, B. *Macromolecules* **1998**, *31*, 5647-5654. (c) Seredyuk, M.; Gaspar, A. B.; Ksenofontov, V.; Galyametdinov, Y.; Verdager, M.; Villain, F.; Gütllich, P. *Inorg. Chem.* **2008**, *47*, 10232-10245. (d) Barberá, J.; Lantero, I.; Moyano, S.; Serrano, J. L.; Elduque, A.; Giménez, R. *Chem. Eur. J.* **2010**, *16*, 14545-14553. (e) Seredyuk, M.; Gaspar, A. B.; Ksenofontov, V.; Galyametdinov, Y.; Verdager, M.; Villain, F.; Gütllich, P. *Inorg. Chem.* **2010**, *49*, 10022-10031. (f) Dechambenoit, P.; Ferlay, S.; Donnio, B.; Guillon, D.; Hosseini, M. W. *Chem. Commun.* **2011**, *47*, 734-736.
- (11) (a) Camerel, F.; Strauch, P.; Antonietti, M.; Faul, C. F. J. *Chem. Eur. J.* **2003**, *9*, 3764-3771. (b) Camerel, F.; Barberá, J.; Otsuki, J.; Tokimoto, T.; Shimazaki, Y.; Chen, L.-Y.; Liu, S.-H.; Lin, M.-S.; Wu, C.-C.; Ziessel, R. *Adv. Mater.* **2008**, *20*, 3462-3467. (c) Li, W.; Yin, S.; Wang, J.; Wu, L. *Chem. Mater.*, **2008**, *20*, 514-522. (d) Molard, Y.; Ledneva, A.; Amela-Cortes, M.; Cîrcu, V.; Naumov, N. G.; Mériade, C.; Artzner, F.; Cordier, S. *Chem. Mater.* **2011**, *23*, 5122-5130. (e) Floquet, S.; Terazzi, E.; Hijazi, A.; Guénée, L.; Piguet, C.; Cadot, E. *New. J. Chem.* **2012**, *36*, 865-868.
- (12) (a) Donnio, B.; Buathong, S.; Bury, I.; Guillon, D. *Chem. Soc. Rev.* **2007**, *36*, 1495-1513. (b) Rosen, B. M.; Wilson, C. J.; Wilson, D. A.; Peterca, M.; Imam, M. R.; Percec, V. *Chem. Rev.* **2009**, *109*, 6275-6540. (c) Donnio, B. *Inorg. Chim. Acta* **2014**, *409*, 53-67.
- (13) Baranoff, E. D.; Voignier, J.; Yasuda, T.; Heitz, V.; Sauvage, J.-P.; Kato, T. *Angew. Chem. Int. Ed.* **2007**, *46*, 4680-4683.
- (14) (a) Aprahamian, I.; Yasuda, T.; Ikeda, T.; Saha, S.; Ditchel, W. R.; Isoda, K.; Kato, T.; Stoddart, J. F. *Angew. Chem. Int. Ed.* **2007**, *46*, 4675-4679. (b) Kidowaki, M.; Nakajima, T.; Araki, J.; Inomata, A.; Ishibashi, H.; Ito, K. *Macromolecules* **2007**, *40*, 6859-6862. (c) Yasuda, T.; Tanabe, K.; Tsuji, T.; Coti, K. K.; Aprahamian, I.; Stoddart, J. F.; Kato, T. *Chem. Commun.* **2010**, *46*, 1224-1226. (d) Suhan, N. D.; Loeb, S. J.; Eichorn, S. H. *J. Am. Chem. Soc.* **2013**, *135*, 400-408.
- (15) Terazzi, E.; Jensen, T. B.; Donnio, B.; Buchwalder, K.; Bourgogne, C.; Rogez, G.; Heinrich, B.; Gallani, J.-L.; Piguet, C. *Dalton Trans.* **2011**, *40*, 12028-12032.
- (16) (a) Lenoble, J.; Maringa, N.; Campidelli, S.; Donnio, B.; Guillon, D.; Deschenaux, R. *Org. Lett.* **2006**, *8*, 1851-1854. (b) J. Lenoble, S. Campidelli, N. Maringa, B. Donnio, D. Guillon, N. Yevlampieva, R. Deschenaux. *J. Am. Chem. Soc.* **2007**, *129*, 9941-9952. (c) Deschenaux, R.; Donnio, B.; Guillon, D. *New. J. Chem.* **2007**, *31*, 1064-1073.
- (17) Barry, N. P. E.; Furrer, J.; Freudenreich, J.; Süß-Fink, G.; Therrien, B. *Eur. J. Inorg. Chem.* **2010**, 725-728.
- (18) (a) Neve, F. *Adv. Mater.* **1996**, *8*, 277-289. (b) Bruce, D. W. *Acc. Chem. Res.* **2000**, *33*, 831-840. (c) Binnemans, K. *Chem. Rev.* **2005**, *105*, 4148-4204. (d) Pucci, D. *Liq. Cryst.* **2011**, *38*, 1451-1465.
- (19) (a) Therrien, B. *Eur. J. Inorg. Chem.* **2009**, 2445-2453. (b) Barry, N. P. E.; Therrien, B. *Eur. J. Inorg. Chem.* **2009**, 4695-4700. (c) Barry, N. P. E.; Zava, O.; Dyson, P. J.; Therrien, B. *Chem. Eur. J.* **2011**, *17*, 9669-9677. (d) Schmitt, F.; Freudenreich, J.; Barry, N. P. E.; Juillierat-Jeanneret, L.; Therrien, B. *J. Am. Chem. Soc.* **2012**, *134*, 754-757.

- (20) (a) Pitto-Barry, A.; Barry, N. P. E.; Zava, O.; Deschenaux, R.; Dyson, P. J.; Therrien, B. *Chem. Eur. J.* **2011**, *17*, 1966-1971. (b) Pitto-Barry, A.; Barry, N. P. E.; Zava, O.; Deschenaux, R.; Therrien, B. *Chem. Asian J.* **2011**, *6*, 1595-1603.
- (21) Moore, J. S.; Stupp, S. I. *Macromolecules* **1990**, *23*, 65-70.
- (22) Dardel, B.; Guillon, D.; Heinrich, B.; Deschenaux, R. *J. Mater. Chem.* **2001**, *11*, 2814-2831.
- (23) Garci, A.; Marti, S.; Schürch, S.; Therrien, B. *RSC Adv.* **2014**, *4*, 8597-8604.
- (24) Barry, N. P. E.; Furrer, J.; Therrien, B. *Helv. Chim. Acta* **2010**, *93*, 1313-1328.
- (25) Campidelli, S.; Lenoble, J.; Barberá, J.; Paolucci, F.; Marcaccio, M.; Paolucci, D.; Deschenaux, R. *Macromolecules* **2005**, *38*, 7915-7925.
- (26) Marcos, M.; Giménez, R.; Serrano, J.-L.; Donnio, B.; Heinrich, B.; Guillon, D. *Chem. Eur. J.* **2001**, *7*, 1006-1013.
- (27) Nagy, Z. T.; Heinrich, B.; Guillon, D.; Tomczyk, J.; Stumpe, J.; Donnio, B. *J. Mater. Chem.* **2012**, *22*, 18614-18622.
- (28) Guillon, D.; Skoulios, A. *Mol. Cryst. Liq. Cryst.* **1983**, *91*, 341-352.
- (29) (a) Hamley, I. W. *Angew. Chem. Int. Ed.* **2003**, *42*, 1692-1712. (b) Tschierske, C. *Annu. Rep. Prog. Chem., Sect. C* **2001**, *97*, 191-267. (c) Lee, M.; Cho, B. K.; Zin, W. C. *Struct. Bond.* **2008**, *128*, 63-98. (d) Tschierske, C. *Angew. Chem. Int. Ed.* **2013**, *52*, 8828-8878.
- (30) (a) Xu, B.; Swager, T. M. *J. Am. Chem. Soc.* **1993**, *115*, 1159-1160. (b) Swager, T. M.; Xu, B. *J. Incl. Phenom. Mol. Recogn. Chem.* **1994**, *19*, 389-398. (c) Felder, D.; Heinrich, B.; Guillon, D.; Nicoud, J.-F.; Nierengarten, J.-F. *Chem. Eur. J.* **2000**, *6*, 3501-3507.
- (31) *OkI: k+l=2n, hhl: l=2n, h00: h=2n. International Tables for Crystallography* (Ed: T. Hahn), 4th ed., Kluwer Academic, Dordrecht, The Netherlands **1995**, 592, Vol. A.
- (32) (a) Diele, S. *Curr. Opin. Colloid Interface Sci.* **2002**, *7*, 333-342. (b) Kutsumizu, S. *Curr. Opin. Solid State Mater. Sci.* **2003**, *6*, 537-543.
- (33) (a) Levelut, A.-M.; Clerc, M. *Liq. Cryst.* **1998**, *24*, 105-115. (b) Impéror-Clerc, M. *Curr. Opin. Colloid Interface Sci.* **2005**, *9*, 370-376.
- (34) La, Y. Park, C. Shin, T. J. Joo, S. H. Kang, S. Kim, K. T. *Nature Chem.* **2014**, *6*, 534-541.
- (35) Zeng, X.; Ungar, G.; Impéror-Clerc, M. *Nature Mater.* **2005**, *4*, 562-567.
- (36) Fogden, A.; Lidin, S. *J. Chem. Soc. Faraday Trans.* **1994**, *90*, 3423-3431.
- (37) The distribution of segregated domains in the cubic cell was analyzed using the following mathematical equation for the isosurfaces (from references 33 and 36):  $\cos(2\pi x) + \cos(2\pi y) + \cos(2\pi z) - \cos(2\pi x)\cos(2\pi y)\cos(2\pi z) = \beta$ , where variables x, y and z refer to the unit cell normed to 1, and  $\beta$  defines the successive isosurfaces, starting from  $\beta = 0$ , for the approximation of the IPMS of type P, up to  $|\beta| = 2$  on edges and vertices. The curvature regularly increases with  $|\beta|$ , while the isosurface area in the normed cell, A, simultaneously shrinks from 2.35 on IPMS to 0 on the edges. Isosurfaces  $\beta$  are hypothesized to coincide with domain boundaries when the volume fractions ( $X_v/2$ ) separating them from the IPMS accord with the real volume of segments. The distribution of interfaces is therefore determined by these volume contributions and by the relation between  $\beta$ ,  $X_v$  and A (Figures S8 and S9). The latter was obtained by numeric integration over cell volume and over membranes going through isosurfaces. The ratio of the isosurface area in the real cell  $a^2 A$  and of the number of molecules per cell  $a^3/V_{mol}$  gives access to areas per segment that can be compared to natural cross-sections.
- (38) Mesogens/aliphatic interfaces ( $[P_2 \subset I][DOS]_4$ :  $X_v \approx 0.31$ ,  $|\beta| \approx 1.44$ ,  $A = 1.92$ ;  $[P_3 \subset I][DOS]_4$ :  $X_v \approx 0.28$ ,  $|\beta| \approx 1.30$ ,  $A = 2.015$ ) and dendritic/aliphatic interfaces ( $[P_2 \subset I][DOS]_4$ :  $X_v \approx 0.44$ ,  $|\beta| \approx 1.89$ ,  $A = 1.21$ ;  $[P_3 \subset I][DOS]_4$ :  $X_v \approx 0.41$ ,  $|\beta| \approx 1.80$ ,  $A = 1.47$ ).